# Selective Oxidation of Oleanane Triterpenoids 

By Yoshimasa Kobayashi, Muneharu Ogawa, and Yukio Ogihara,* Faculty of Pharmaceutical Sciences,<br>Nagoya City University, Tanabe-dori, Mizuhoku, Nagoya 467, Japan


#### Abstract

Reagent A (exact composition unknown) was obtained from a mixture of $\mathrm{CrO}_{3}$, pyridine, and BunOH saturated with water. The reaction of the oleanane triterpenoid glycosides (1), (4), and (5) with reagent A gave 11-oxocompounds (3) and (6), which indicated that this reagent is useful for the oxidation of an allyl alcohol and an allyl ether. On the other hand, the reaction of oleanane triterpenoids (11), (13), (14), and (15) with reagent A gave 16 -oxo-derivatives (9), (16), (17), and (18), while compounds (19) and (22) with this reagent gave 21 -oxoderivatives (24) and (25). Reagent A is also very useful for the selective oxidation of 16-and 21 -axial hydroxygroups.


A previous paper of ours ${ }^{\mathbf{1}}$ has described the oxidation of the glycoside (1) with chromium trioxide-pyridine followad by acidic hydrolysis to give the diene (2) and $3,11,21-$ trioxo-olean-12-en-28-oic acid (3). It seemed strange that the methoxy-group was oxidized to a carbonyl group by chromic trioxide-pyridine. However a model

(2)
experiment revealed the formation of the carbonyl group during the acidic treatment of the $n$-butanolic extract of the reaction mixture. This paper describes the preparation of the reagent A , suggested by the above results, and its application to some oleanane triterpenoids.

## RESULTS AND DISCUSSION

Preparation of Reagent $A$.-A mixture of $\mathrm{CrO}_{3}$, pyridine and $\mathrm{Bu}^{\mathrm{n}} \mathrm{OH}$ saturated with water were kept at room temperature for a day. The mixture was concenfrated under reduced pressure and the resultant precipitate was filtered off, washed with water and $\mathrm{CHCl}_{3}$, and dried under reduced pressure to afford a brown powder, termed reagent A .

Some Reactions on the Oleanane Triterpenoids.-The reaction of the glycoside (1) with reagent A followed by acidic treatment gave the diene (2) and 3,11,21-trioxo-olean-12-en-28-oic acid (3). On the other hand, the reaction of saikosaponin a (4) ${ }^{2}$ and $b_{3}(5)^{3}$ with an allyl ether group with reagent $A$ gave the same ll-ketoderivative (6), the structure of which was determined on the basis of its ${ }^{1} \mathrm{H}$ n.m.r. spectrum [ $\mathbf{\delta} 5.56(1 \mathrm{H}, \mathrm{s}, \mathbf{1 2 - H})$ ]. From these experiments, reagent $A$ was found to be useful for the oxidation of an allyl alcohol or an allyl

Saikosaponin a
(4)

(6) $R=H$
(6a) $R=A c$

Saikosaponin $b_{3}$
(5)


(13)
(16) $R=H$
(14)
(17)
(16a) $R=A C$

(18)

ether group to an $\alpha \beta$-unsaturated ketone without affecting the other hydroxy-groups. Rotundioside F (7) ${ }^{4}$ (isolated from the leaves of Bupleurivm rotundifolium L.), which has an allyl ether group, when treated with reagent A gave not the expected 11-oxo-olean-12-en- $3 \beta$,$16 \alpha, 28$-triol (8) but the 16-oxo-oleana-11,13(18)-diene$3 \beta, 28$-diol (9). The structure (9) was deduced from the following chemical and spectroscopic evidence. Compound (9), $\mathrm{C}_{30} \mathrm{H}_{46} \mathrm{O}_{3}$, exhibits a broad OH absorption and sharp band at $1688 \mathrm{~cm}^{-1}$, indicative of a carbonyl group, in the i.r. spectrum, and ${ }^{1} \mathrm{H}$ n.m.r. signals for two olefinic protons (AB quartet, $\delta 5.71$ and $6.50, J 11 \mathrm{~Hz}$ ) on a disubstituted double bond and three easily identified protons next to oxygens ( $3 \alpha-\mathrm{H}: \delta 3.22$, dd, $J 9$ and 8 Hz ; $28-\mathrm{CH}_{2}: \delta 3.63$ and $3.95, \mathrm{AB}$ quartet, $J 11 \mathrm{~Hz}$ ). Compound (9) was then treated with acetic anhydridepyridine to give the diacetate (10), ${ }^{5}$ which was identified by comparison with an authentic sample prepared from rotundiogenin $\mathrm{B}(11)^{4}$ via rotundiogenin B diacetate (12). It is interesting that the selective oxidation of the $16 \alpha-$ hydroxy-group (axial) was achieved without affecting the other hydroxy-groups. The reaction of other oleanane triterpenoids with $16 \alpha$-hydroxy-group, i.e. compounds (11), (13), (14), and (15), with reagent A were then examined [solvent, dioxan- $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}(\mathbf{l}: 1)$; reflux for $30 \mathrm{~min}]$. Rotundiogenin $\mathrm{B}(11)$ gave a $90 \%$ yield of 16 -oxo-oleana-11,13(18)-diene- $3 \beta, 28$-diol (9) as a single product. Echinocystic acid (13), isolated from the fresh leaves of Bupleurum rotundifolium L., reacted with reagent A to afford a mixture of two stereoisomeric 16 -keto-derivatives (16), which were isomeric at $\mathrm{C}-17$ based on g.l.c. analysis of the acetates (16a) ${ }^{6}(\alpha: \beta=1: 3)$. Primulagenin A (14), derived from echinocystic acid (13), was treated with reagent $A$ to give the 16 -keto-derivative (17) only. The dihydro-rotundiogenin B(15), derived by catalytic reduction of rotundiogenin $B$ (11), was treated with reagent A to give an $85 \%$ yield of 16 -oxo-olean-$13(18)$-ene- $3 \beta, 28$-diol (18) as a single product. Thus the $16 \alpha$-hydroxy-group (axial) is selectively oxidized to a ketone function in some oleanane triterpenoids. On the other hand, when the known papyriogenin $\mathrm{A}(2)$ and C (21) were reduced with lithium aluminium hydride, ${ }^{7}$ compounds (19) and (20) $[3 \beta, 21 \beta, 28$ - and $3 \beta, 21 \alpha, 28$-trihydr-oxyoleana-11,13(18)-diene] were obtained from (2) and compounds (22) and (23) $[3 \alpha, 21 \beta, 28$ - and $3 \alpha, 21 \alpha, 28$-tri-hydroxyoleana-11,13(18)-diene] were obtained from (21). Compounds (19) and (22) have a $21 \alpha$-axial hydroxygroup, compounds (22) and (23) a $3 \alpha$-axial hydroxygroup. The reaction of reagent A with compounds (19), (20), (22), and (23) was then examined to clarify the selective oxidation of axial hydroxy-groups at C-3 and C-21. Reagent A with compounds (19) and (22) gave 21-oxo-derivatives (24) and (25), but compounds (20) and (23) were recovered unchanged. The structures of the 21-oxo-derivatives (24) and (25) were assigned on the basis of ${ }^{1} \mathrm{H}$ n.m.r. analysis. This indicated that reagent A was also useful for the selective oxidation of a 21 -axial-hydroxy-group. The selective oxidation of the 16 - and 21-axial hydroxy-groups with reagent $A$ seems to be in-
fluenced by the presence of a C-28 hydroxy-group. To a solution of rotundioside $\mathrm{E}(26)^{4}$ in dioxan- $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}$ was added reagent $A$ and the total mixture was stirred at room temperature for 15 h . This reaction gave a $79 \%$


| R | R1 |
| :---: | :---: |
| (2) $=0$ | $=0$ |
| (21) $\cdots \mathrm{OH}(\mathrm{ax})$ | $=0$ |
| (19) $-\mathrm{OH}(\mathrm{eq})$ | $-\mathrm{OH}(\mathrm{ax})$ |
| (20) $-O H(e q)$ | $--\mathrm{OH}(\mathrm{eq})$ |
| (22) $\cdots$ OH(ax) | $-\mathrm{OH}(\mathrm{ax})$ |
| (23) $\cdots$ - $\mathrm{OH}(\mathrm{ax}$ ) | ---OH(eq) |


yield of the 28-hydroxy-16-oxo-oleana-11,13(18)-dien$3 \beta$-yl $\alpha$-L-rhamnopyranosyl-( $1 \longrightarrow 2$ )- $\beta$-D-glucopyranosyl$(1 \longrightarrow 2)$ - $\beta$-D-fucopyranoside (27), the structure of which was confirmed mainly by ${ }^{13} \mathrm{C}$ n.m.r. analysis. The ${ }^{13} \mathrm{C}$ n.m.r. signal for C-16 at $\delta 67.7$ of compound (26) shifted to low field [ $\delta 214.8$ in (27)] due to carbonyl formation and a complete identity in the sugar moieties of both compounds was observed. The signal assignments were fairly straightforward by comparison of its ${ }^{13} \mathrm{C}$ n.m.r. spectrum with that of the mother compound (9) and literature data of saikogenins ${ }^{8,9}$ (Table).

|  | (11) ${ }^{a}$ | (11a) | (26) ${ }^{a}$ | (27) | (9) | (10) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-1 | $39.0{ }^{\text {b }}$ | (37.9) | $39.1{ }^{\text {b }}$ | $38.4{ }^{\text {b }}$ | $38.5{ }^{\text {b }}$ | (37.9) |
| 2 | 28.1 | (23.4) | 26.6 | 26.4 | 27.9 | (23.4) |
| 3 | 78.0 | (80.7) | 89.6 | 89.3 | 77.9 | (80.6) |
| 4 | 39.5 | (37.9) | 39.9 | 39.7 | 39.4 | (37.9) |
| 5 | 55.4 | (55.0) | 55.6 | 54.2 | 54.3 | (54.8) |
| 6 | 18.9 | (18.2) | 18.9 | 18.4 | 18.6 | (18.1) |
| 7 | $32.6{ }^{\text {c }}$ | (32.2) | $32.6{ }^{\text {c }}$ | 32.8 | 32.8 | (32.1) |
| 8 | $41.1{ }^{\text {d }}$ | (41.0) | $41.1{ }^{\text {d }}$ | 40.2 | 40.3 | (40.0) |
| 9 | 53.9 | (53.4) | 53.9 | 53.8 | 54.0 | (53.4) |
| 10 | 37.0 | (36.5) | 36.6 | 36.6 | 37.0 | (36.6) |
| 11 | 126.2 | (126.9) | 126.2 | 127.5 | 127.5 | (127.6) |
| 12 | 126.2 | (125.3) | 126.2 | 124.8 | 124.9 | (124.5) |
| 13 | 136.1 | (136.8) | 136.0 | 136.5 | 136.6 | (136.6) |
| 14 | $41.9{ }^{\text {d }}$ | (41.4) | $41.9{ }^{\text {d }}$ | 47.3 | 47.3 | (46.9) |
| 15 | $31.9{ }^{\text {c }}$ | (28.7) | $31.9{ }^{\text {c }}$ | 45.6 | 45.7 | (44.7) |
| 16 | 67.7 | (71.1) | 67.7 | 214.8 | 214.7 | (213.9) |
| 17 | 45.3 | (40.9) | 45.3 | 55.1 | 55.0 | (50.8) |
| 18 | 133.1 | (129.3) | 133.1 | 133.3 | 133.4 | (131.2) |
| 19 | $38.6{ }^{\text {b }}$ | (38.3) | $38.5{ }^{\text {b }}$ | $38.2{ }^{\text {b }}$ | $38.2{ }^{\text {b }}$ | (37.9) |
| 20 | 32.6 | (32.2) | 32.6 | 33.4 | 33.4 | (33.5) |
| 21 | 35.5 | (34.4) | 35.6 | 35.2 | 35.2 | (34.9) |
| 22 | 24.5 | (24.3) | 24.5 | 27.4 | 27.4 | (27.2) |
| 23 | 28.5 | (27.8) | 28.2 | 28.0 | 28.5 | (27.8) |
| 24 | 16.0 | (16.1) | 16.3 | 16.2 | 16.0 | (16.2) |
| 25 | $18.4{ }^{\text {e }}$ | (18.2) | $18.3{ }^{e}$ | $18.0{ }^{e}$ | $18.1{ }^{\text {e }}$ | (18.0) |
| 26 | $17.3{ }^{\text {e }}$ | (16.9) | $17.4{ }^{e}$ | $17.1{ }^{e}$ | $17.2{ }^{e}$ | (17.0) |
| 27 | 21.9 | (20.9) | 21.8 | 22.0 | 22.1 | (21.8) |
| 28 | 64.7 | (65.4) | 64.8 | 65.8 | 65.8 | (65.9) |
| 29 | 25.1 | (24.6) | 25.1 | 24.0 | 24.0 | (23.8) |
| 30 | 32.6 | (32.2) | 32.6 | 32.2 | 32.2 | (32.5) |
|  |  |  | 105.3 | 105.1 |  |  |
| 2 |  |  | 78.0 | 78.0 |  |  |
| Fucose 3 |  |  | 76.2 | 76.1 |  |  |
| Fucose 4 |  |  | 72.8 | 72.8 |  |  |
| 5 |  |  | 70.9 | 70.8 |  |  |
| 6 |  |  | 17.4 | 17.3 |  |  |
| 1 |  |  | $101.8{ }^{f}$ | $101.7{ }^{f}$ |  |  |
| 2 |  |  | 79.5 | 79.3 |  |  |
| Glucose 3 |  |  | 77.2 | 76.9 |  |  |
| Glucose 4 |  |  | 72.8 | 72.7 |  |  |
| 5 |  |  | 77.2 | 77.2 |  |  |
| 6 |  |  | 63.3 | 63.3 |  |  |
| 1 |  |  | $102.2{ }^{\text {f }}$ | $102.0{ }^{f}$ |  |  |
| 2 |  |  | 72.8 | 72.4 |  |  |
| Rhamnose 3 |  |  | 72.8 | 72.7 |  |  |
| $4$ |  |  | 74.3 | 74.2 |  |  |
| 5 |  |  | 69.5 | 69.4 |  |  |
| 6 |  |  | 18.7 | 18.9 |  |  |

${ }^{a}$ The assignments of these signals have been revised: cf. ref. 4. $\quad b-f$ Assignments may be reversed in each vertical column. Peracetates of (9) and (11) (in parentheses) in $\mathrm{CDCl}_{3}$.

In conclusion, reagent A is found to be very useful for the selective oxidation of an allyl alcohol, an allyl ether, and 16 - and 21 -axial hydroxy-groups. The composition of this reagent is under investigation.

## EXPERIMENTAL

M.p.s were measured with a Yanagimoto micro-apparatus. Unless otherwise stated, optical rotation were neasured on solutions in chloroform, i.r. spectra on KBr discs, and ${ }^{1} \mathrm{H}$ n.m.r. spectra on solutions in deuteriochloroform. ${ }^{1} \mathrm{H}$ N.m.r. spectra were recorded using a JEOL Model JNM-MH-100 spectrometer, employing $\mathrm{SiMe}_{4}$ as an internal standard, and ${ }^{13} \mathrm{C}$ n.m.r. spectra on a JEOL Model JNM-FX-100 spectrometer on $\left[{ }^{2} \mathrm{H}_{5}\right]$ pyridine solutions containing SiMe $_{4}$ as internal reference, in $5-\mathrm{mm}$ spinning tubes at room temperature.

Preparation of Reagent $A$.-A mixture of $\mathrm{CrO}_{3}$ ( $500 \mathrm{~m} g$ ), pyridine ( 30 ml ), and n-butyl alcohol saturated with water $(250 \mathrm{ml})$ was kept at room temperature for one day. The mixture was concentrated under reduced pressure and the
residue washed with water and chloroform, then dried under reduced pressure to give the reagent $A(520 \mathrm{mg})$ as a brown powder. Reagent $A$ is soluble only in sulphuric acid, and is insoluble in acetone, chloroform, pyridine, ether, and water; $\nu_{\max }(\mathrm{KBr}) 3380,1607,1485,1445,1218,1153$, $1068,1045,1015,905,800,693,640$, and $510 \mathrm{~cm}^{-1}$; $\lambda_{\text {max }}$ $\left(2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}\right.$ ) 255.5 ( $\varepsilon 10400$ ) nm (Found: C, 26.55 ; H, 2.9 ; $\mathrm{N}, 6.05 ; \mathrm{O}, 17.9 . \quad \mathrm{C}_{10} \mathrm{H}_{14} \mathrm{Cr}_{4} \mathrm{~N}_{2} \mathrm{O}_{5}$ requires C , 26.65; H , 3.1 ; N, 6.2 ; O, $17.8 \%$ ).

Reaction of the Oleanane Triterpenoid Glycosides (1), (4), and (5) with Reagent $A$.-(a) To a stirred solution of reagent A ( 200 mg ) in dioxan ( 2 ml ) and $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}(2 \mathrm{ml})$, papyrioside L-IIa ( 1200 mg ) in dioxan ( 1 ml ) was added. The mixture was refluxed for 30 min , then diluted with water and extracted with ether. The ether layer was washed with water and evaporated and the residue was chromatographed on silica gel to give papyriogenin $A(40 \mathrm{mg})$ and 3,11,21-trioxo-olean-12-en-28-oic acid (3) (90 mg), m.p. $25 \mathrm{l}-254{ }^{\circ} \mathrm{C}$ (Found: C, 76.85; H, 9.55. $\mathrm{C}_{30} \mathrm{H}_{44} \mathrm{O}_{4}$ requires $\mathrm{C}, 76.9 ; \mathrm{H}, 9.45 \%$ ), identical with an authentic specimen ${ }^{1}$ (mixed m.p., t.l.c., and ${ }^{1} \mathrm{H}$ n.m.r. spectrum).
(b) Saikosaponin a (4) ( 70 mg ) and saikosaponin $\mathrm{b}_{3}$ (5) $(100 \mathrm{mg})$ were worked up as in $(a)$ and the reaction mixture was purified by preparative t.l.c. to afford 11-oxo-oleana~12-en- $3 \beta, 16 \beta, 23,28$-tetraol (6) ( 12 mg ), not crystalline; $\delta$ $\left(\mathrm{CDCl}_{3}-\mathrm{CD}_{3} \mathrm{OD}\right) 0.74(3 \mathrm{H}, \mathrm{s}), 0.96(6 \mathrm{H}, \mathrm{s}), 1.19(6 \mathrm{H}, \mathrm{s})$, $1.50(3 \mathrm{H}, \mathrm{s}), 3.20-3.88(6 \mathrm{H}, \mathrm{m})$, and $5.56(1 \mathrm{H}, \mathrm{s})$. Compound (6) ( 10 mg ) was acetylated in the usual way to afford the tetra-acetate ( 6 a ) ( 12 mg ), not crystalline; $\delta 0.87$ $(3 \mathrm{H}, \mathrm{s}), 0.96(6 \mathrm{H}, \mathrm{s}), 1.20(6 \mathrm{H}, \mathrm{s}), 1.51(3 \mathrm{H}, \mathrm{s}), 3.72(1 \mathrm{H}, \mathrm{d}$, $J 11 \mathrm{~Hz}), 3.92(2 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 4.18(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 4.82$ $(1 \mathrm{H}, \mathrm{t}, J 8 \mathrm{~Hz}), 5.60(1 \mathrm{H}, \mathrm{dd}, J 10$ and 6 Hz$)$, and $5.65(1 \mathrm{H}$, s).

Reaction of Rotundioside $F(7)$ with Reagent $A$.-Rotundioside $F(7)(250 \mathrm{mg})$ was treated with reagent $A$ as above and the reaction mixture was purified by column chromatography to give compound (9) ( $80 \mathrm{mg}, 64 \%$ ) as colourless needles from benzene, m.p. $213-216^{\circ} \mathrm{C},[\alpha]_{\mathrm{p}}-15.5^{\circ}(c 0.32)$; $\nu_{\text {max. }}$ 3450 and $1688 \mathrm{~cm}^{-1}$; m/e $454\left(M^{+}\right)$; $\delta\left(\mathrm{CDCl}_{3}-\mathrm{CD}_{3} \mathrm{OD}\right)$ $0.76(3 \mathrm{H}, \mathrm{s}), 0.79(3 \mathrm{H}, \mathrm{s}), 0.92(3 \mathrm{H}, \mathrm{s}), 0.94(3 \mathrm{H}, \mathrm{s}), 0.95$ $(3 \mathrm{H}, \mathrm{s}), 1.01(6 \mathrm{H}, \mathrm{s}), 3.22(1 \mathrm{H}, \mathrm{t}, J 8 \mathrm{~Hz}), 3.63(1 \mathrm{H}, \mathrm{d}, J 11$ $\mathrm{Hz}), 3.95(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 5.71(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, and 6.50 ( $1 \mathrm{H}, \mathrm{dd}, J 11$ and 3 Hz ) (Found: $\mathrm{C}, 77.7 ; \mathrm{H}, 10.1 . \mathrm{C}_{30} \mathrm{H}_{46}{ }^{-}$ $\mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 77.7 ; \mathrm{H}, 10.2 \%$ ); compound (9) (20 mg ) was acetylated in the usual way to give the diacetate (10) (22 mg), m.p. $247-250^{\circ} \mathrm{C} ; \delta 3.93(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, $4.53(1 \mathrm{H}, \mathrm{dd}, J 9$ and 8 Hz$), 4.70(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 5.72$ ( $1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}$ ), and $6.49(1 \mathrm{H}, \mathrm{dd}, J 10$ and 3 Hz ) (Found: $\mathrm{C}, 75.65 ; \mathrm{H}, 9.5 . \quad \mathrm{C}_{34} \mathrm{H}_{50} \mathrm{O}_{5}$ requires $\mathrm{C}, 75.8 ; \mathrm{H}, 9.35 \%$ ), identical with an authentic specimen ${ }^{10}$ (mixed m.p., and i.r. and n.m.r. spectra).

Reaction of the Oleanane Tviterpenoids (11) and (13)-(15) with reagent $A$.-(a) Rotundiogenin $B(11)(100 \mathrm{mg})$ and reagent $\mathrm{A}(100 \mathrm{mg})$ were dissolved in dioxan $(2 \mathrm{ml})$ and $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}(2 \mathrm{ml})$, and refluxed for 30 min . The solution was extracted with ether and treated as usual to afford 16 -oxo-oleana-11,13(18)-diene- $3 \beta$, 28 -diol ( 9 ) ( $91 \mathrm{mg}, 90 \%$ ), identical with compound (9) obtained as above (t.l.c. and n.m.r. spectrum).
(b) Echinocystic acicl (13) (92 mg) was worked up as in (a) and the product was chromatographed on silica gel [benzeneacetone $(10: 1)]$ to afford $3 \beta$-hydroxy-16-oxo- 28 -norolean12 -ene ${ }^{6}$ ( 16 ) (maragenin I) ( $40 \mathrm{mg}, 48 \%$ ). Compound ( 16 ) was recrystallized from methanol, m.p. $207-209{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}$ $+32.1^{\circ}(c 0.12) ; \nu_{\max .} 3470$ and $1693 \mathrm{~cm}^{-1} ; m / e 426\left(M^{+}\right)$,

218, and 190; $\delta 3.20(1 \mathrm{H}, \mathrm{t}, J 8 \mathrm{~Hz})$, and $5.44(1 \mathrm{H}$, pseudotriplet, $J 4 \mathrm{~Hz}$ ) (Found: C, 81.85; H, 10.9. $\mathrm{C}_{29} \mathrm{H}_{46} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.65 ; 10.85 \%$ ). Acetylation of compound (16) $(34 \mathrm{mg})$ gave a stereoisomeric mixture of the monoacetates (16a) ( 38 mg ). The monoacetates (16a) were not crystalline; $\delta 2.08(3 \mathrm{H}, \mathrm{s}), 4.56(1 \mathrm{H}, \mathrm{t}, J 8 \mathrm{~Hz})$ and $5.54(1 \mathrm{H}, \mathrm{t}$, $J 4 \mathrm{~Hz})$; g.l.c. [ $0.5 \% \mathrm{SE}-30$ on Chromosorb W ( $60-80$ mesh) $]$ showed the presence of the $17 \alpha$ - and $17 \beta$-isomers.
(c) Primulagenin A (14) $(50 \mathrm{mg})$ was worked up as in (a) and the product was chromatographed on silica gel $\left[\mathrm{CHCl}_{3}-\right.$ MeOH ( $30: 1$ )] to afford 16-oxo-oleana-12-ene- $3 \beta, 28$-diol (17) ( $29 \mathrm{mg}, 58 \%$ ) as a colourless powder (from $\mathrm{MeOH}-$ $\mathrm{H}_{2} \mathrm{O}$ ), m.p. $211-213{ }^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}+18.2^{\circ}(c 0.22) ; \nu_{\max } 3410$ and $1698 \mathrm{~cm}^{-1} ; m / e 456\left(M^{+}\right), 248$ and $208 ; \delta 0.80(3 \mathrm{H}, \mathrm{s})$, $0.88(6 \mathrm{H}, \mathrm{s}), 0.95(3 \mathrm{H}, \mathrm{s}), 1.00(3 \mathrm{H}, \mathrm{s}), 1.03(3 \mathrm{H}, \mathrm{s}), 1.22$ $(3 \mathrm{H}, \mathrm{s}), 3.19(1 \mathrm{H}, \mathrm{dd}, J 8$ and 9 Hz$), 3.40(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, $3.94(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, and $5.48\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, W_{\frac{1}{2}} 7 \mathrm{~Hz}\right)$ (Found: $\mathrm{C}, 77.2 ; \mathrm{H}, 10.5 . \mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires C , 77.35 ; H, 10.6\%).
(d) Synthesis of dihydrorotundiogenin $B$ (15). Rotundiogenin B ( 150 mg ) was dissolved in glacial acetic acid ( 5 ml ) and shaken with $\mathrm{PtO}_{2}(100 \mathrm{mg})$ under an atmosphere of $\mathrm{H}_{2}$ for 2 h . The catalyst was filtered off and the filtrate was evaporated in vacuo. Recrystallization of the residue from $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ afforded as a colourless powder ( 120 mg ) the dihydro-derivative, olean-13(18)-ene- $3 \beta, 16 \alpha$-28-triol (15), m.p. $228-231{ }^{\circ} \mathrm{C}$; $\delta\left(\mathrm{CDCl}_{3}-\mathrm{CD}_{3} \mathrm{OD}\right) 0.77(6 \mathrm{H}, \mathrm{s}), 0.86$ $(3 \mathrm{H}, \mathrm{s}), 0.97(9 \mathrm{H}, \mathrm{s}), 1.24(3 \mathrm{H}, \mathrm{s}), 3.14(1 \mathrm{H}, \mathrm{t}, J 8 \mathrm{~Hz}), 3.40$ $(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 3.90(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, and $3.96(1 \mathrm{H}, \mathrm{t}$, $J 3 \mathrm{~Hz}$ ) (Found: C, 75.3; H, 11.0. $\quad \mathrm{C}_{30} \mathrm{H}_{50} \mathrm{O}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$ requires C, $75.6 ; \mathrm{H}, 11.0 \%$ ). This was acetylated with acetic anhydride-pyridine to give the triacetate; $\delta 0.80(3 \mathrm{H}, \mathrm{s})$, $0.84(3 \mathrm{H}, \mathrm{s}), 0.87(6 \mathrm{H}, \mathrm{s}), 0.90(3 \mathrm{H}, \mathrm{s}), 0.96(3 \mathrm{H}, \mathrm{s}), 1.26$ $(3 \mathrm{H}, \mathrm{s}), 2.06(9 \mathrm{H}, \mathrm{s}), 3.96(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz}), 4.23(1 \mathrm{H}, \mathrm{d}, J$ $11 \mathrm{~Hz}), 4.49(1 \mathrm{H}, \mathrm{dd}, J 9$ and 8 Hz$)$, and $5.12(1 \mathrm{H}, \mathrm{t}, J 3$ Hz ).

Dihydrorotundiogenin $B(15)(80 \mathrm{mg})$ was worked up as in (a) and the product was chromatographed on silica gel to afford 16 -oxo-olean-13(18)-ene-3 3 -28-diol ( 18 ) ( $68 \mathrm{mg}, 85 \%$ ) as colourless crystals from benzene, m.p. $227-230^{\circ} \mathrm{C},[\alpha]_{\mathrm{D}}$ $-91.7^{\circ}(c 0.12) ; \nu_{\max } 3450$ and $1685 \mathrm{~cm}^{-1} ; m / e 426,424$, and $203(100 \%) ; \delta 0.71(3 \mathrm{H}, \mathrm{s}), 0.76(3 \mathrm{H}, \mathrm{s}), 0.89(3 \mathrm{H}, \mathrm{s})$, $0.96(3 \mathrm{H}, \mathrm{s}), 0.98(3 \mathrm{H}, \mathrm{s}), 1.00(3 \mathrm{H}, \mathrm{s}), 1.11(3 \mathrm{H}, \mathrm{s}), 3.19$ $(1 \mathrm{H}, \mathrm{dd}, J 9$ and 8 Hz$), 3.63(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, and $3.85(1 \mathrm{H}$, d, $J 11 \mathrm{~Hz}$ ) (Found: C, 78.65; H, 10.9. $\mathrm{C}_{30} \mathrm{H}_{48} \mathrm{O}_{3}$ requires C, $78.9 ; \mathrm{H}, \mathbf{1 0 . 6} \%$ ).

Reaction of the Oleanane Triterpenoids (19), (20), (22), and (23) with Reagent A.-(a) Compound (19) ( 53 mg ), derived from papyriogenin $A$ (2) by reduction with lithium aluminium hydride, and reagent $\mathrm{A}(60 \mathrm{mg})$ were dissolved in dioxan ( 2 ml ) and $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}(2 \mathrm{ml})$, and refluxed for 1 h . The solution was extracted with ether, treated as usual, and the product purified by preparative t.1.c. $\left[\mathrm{CHCl}_{3}-\mathrm{MeOH}\right.$ ( $10: 1$ )] to afford 21 -oxo-oleana-11,13(18)-diene-3 $\beta, 28$-diol (24) ( $25 \mathrm{mg}, 47 \%$ ) as colourless crystals from MeOH , m.p. $250-251{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}-37.2^{\circ}(c 0.54)$; $\nu_{\max .}(\mathrm{KBr}) 3420$ and $1704 \mathrm{~cm}^{-1} ; \delta 0.75(3 \mathrm{H}, \mathrm{s}), 0.79(3 \mathrm{H}, \mathrm{s}), 0.92(3 \mathrm{H}, \mathrm{s}), 1.01$ $(3 \mathrm{H}, \mathrm{s}), 1.08(3 \mathrm{H}, \mathrm{s}), 1.10(3 \mathrm{H}, \mathrm{s}), 3.26(1 \mathrm{H}, \mathrm{dd}, J 8$ and 7 $\mathrm{Hz}), 3.52(2 \mathrm{H}, \mathrm{s}), 5.72(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz})$, and $6.52(1 \mathrm{H}, \mathrm{dd}$, $J 10$ and 3 Hz ) (Found: C, 78.1; H, 10.25. $\mathrm{C}_{30} \mathrm{H}_{46} \mathrm{O}_{3} \cdot 0.5$ $\mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 77.7 ; \mathrm{H}, 10.2 \%$ ). It was acetylated in the
usual way to give the diacetate; $[\alpha]_{\mathrm{D}}-65.0^{\circ}(c 0.40) ; \delta 0.72$ $(3 \mathrm{H}, \mathrm{s}), 0.89(6 \mathrm{H}, \mathrm{s}), 0.96(3 \mathrm{H}, \mathrm{s}), 1.04(3 \mathrm{H}, \mathrm{s}), 1.08(3 \mathrm{H}$, s), $1.12(3 \mathrm{H}, \mathrm{s}), 2.08(6 \mathrm{H}, \mathrm{s}), 3.98(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}), 4.18$ $(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}), 4.57(1 \mathrm{H}, \mathrm{dd}, J 8$ and 7 Hz$), 5.79(1 \mathrm{H}, \mathrm{d}$, $J 10 \mathrm{~Hz}$ ), and $6.53(1 \mathrm{H}, \mathrm{dd}, J 10$ and 3 Hz$)$.
(b) Compound (22) ( 92 mg ), derived from papyriogenin C (21) by reduction with lithium aluminium hydride, and reagent A ( 100 mg ) were worked up as in (a) and the product chromatographed on silica gel [benzene-acetone (15:1)] to give 21 -oxo-oleana-11,13(18)-diene- $3 \alpha, 28$-diol (25) ( 44 mg , $49 \%$ ) as colourless needles from $\mathrm{MeOH}, \mathrm{m} . \mathrm{p} .246-248{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}-26.8^{\circ}(c 0.82) ; \nu_{\text {max }}(\mathrm{KBr}) 3430$ and $1700 \mathrm{~cm}^{-1} ; \delta$ $\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right) 0.82(3 \mathrm{H}, \mathrm{s}), 0.90(3 \mathrm{H}, \mathrm{s}), 0.99(6 \mathrm{H}, \mathrm{s}), 1.11(3 \mathrm{H}$, s), $1.16(3 \mathrm{H}, \mathrm{s}), 1.18(3 \mathrm{H}, \mathrm{s}), 3.63\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, W_{\frac{1}{2}} 7 \mathrm{~Hz}\right)$, $3.80(2 \mathrm{H}, \mathrm{s}), 5.87(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz})$, and $6.65(1 \mathrm{H}, \mathrm{dd}, J 10$ and 3 Hz ) (Found: $\mathrm{C}, 78.2 ; \mathrm{H}, 10.55 . \mathrm{C}_{30} \mathrm{H}_{46} \mathrm{O}_{3} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$ requires $\mathrm{C}, 77.7 ; \mathrm{H}, 10.2 \%$ ). It was acetylated in the usual way to give the diacetate; $\delta 0.78(3 \mathrm{H}, \mathrm{s}), 0.88(3 \mathrm{H}$, s), $0.91(3 \mathrm{H}, \mathrm{s}), 0.95(3 \mathrm{H}, \mathrm{s}), 1.09(3 \mathrm{H}, \mathrm{s}), 1.12(3 \mathrm{H}, \mathrm{s}), 1.28$ $(3 \mathrm{H}, \mathrm{s}), 2.08(3 \mathrm{H}, \mathrm{s}), 2.11(3 \mathrm{H}, \mathrm{s}), 3.94(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, $4.16(1 \mathrm{H}, \mathrm{d}, J 11 \mathrm{~Hz})$, and $4.72\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, W_{\frac{1}{2}} 7 \mathrm{~Hz}\right)$.
(c) Compound (20) ( 54 mg ) was worked up as in (a); the product was identical with starting material (t.l.c. and n.m.r. spectrum).
(d) Compound (23) ( 150 mg ) was worked up as in (a); the product was identical with starting material (t.l.c., i.r. and n.m.r. spectra).

Reaction of Rotundioside E (26) with Reagent A at Room Temperature.-Rotundioside $E(26,66 \mathrm{mg})$ and reagent $A$ ( 70 mg ) were dissolved in dioxan ( 3 ml ) and $2 \mathrm{~N}-\mathrm{H}_{2} \mathrm{SO}_{4}(2 \mathrm{ml})$ and stirred at room temperature for 15 h . The solution was extracted with n-butyl alcohol and treated as usual to afford 16 -oxo-rotundioside E (27) ( $52 \mathrm{mg}, 79 \%$ ). Compound (27) was recrystallized from aqueous methanol as a white powder, m.p. $240-243{ }^{\circ} \mathrm{C}$, $[\alpha]_{\mathrm{D}}-67.0^{\circ}$ (c 1.0 in pyridine) ; $\delta\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right) 6.63(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}), 5.61(1 \mathrm{H}, \mathrm{d}, J$ $10 \mathrm{~Hz}), 6.31(1 \mathrm{H}, \mathrm{br}$ s, anomeric proton of rhamnose), 4.52 $(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}$, anomeric proton of glucose), and $4.00(1 \mathrm{H}$, d, $J 7 \mathrm{~Hz}$, anomeric proton of fucose) (Found: C, $63.55 ; \mathrm{H}$, 8.4. $\mathrm{C}_{48} \mathrm{H}_{76} \mathrm{O}_{16}$ requires $\mathrm{C}, 63.4 ; \mathrm{H}, 8.45 \%$ ).

We thank Professor S. Nishibe and Mr. A. Sakushima, Higashi Nihon University, for mass spectra.
[0/1920 Received, 12th December, 1980]

## REFERENCES

${ }^{1}$ M. Takai, S. Amagaya, and Y. Ogihara, J. Chem. Soc., Perkin Trans. 1, 1977, 1801.
${ }_{2}$ T. Kubota and H. Hinoh, Tetrahedron Lett., 1968, 303.
${ }^{3}$ A. Shimaoka, S. Seo, and H. Minato, J. Chem. Soc., Perkin Trans. 1, 1975, 2043.
${ }_{4}$ Y. Kobayashi, T. Takeda, and Y. Ogihara, Chem. Pharm. Bull., in the press.
${ }_{5}$ T. Kubota, F. Tonami, and H. Hinoh, Tetrahedron, 1967, 23, 3333.
${ }^{6}$ P. J. Hylands and A. M. Salama, Tetrahedron, 1979, 35, 417.
${ }^{7}$ M. Asada, S. Amagaya, M. Takai, and Y. Ogihara, J. Chem. Soc., Perkin Trans. 1, 1980, 325.
${ }^{8}$ K. Tori, Y. Yoshimura, S. Seo, K. Sakurai, Y. Tomita, and H. Ishii, Tetrahedron Lett., 1976, 4163.
${ }_{9}$ H. Ishii, M. Nakamura, S. Seo, K. Tori, T. Tozyo, and Y. Yoshimura, Chem. Pharm. Bull., 1980, 28, 2367.
${ }_{10}$ Y. Kobayashi and Y. Ogihara, Chem. Pharm. Bull., in the press.

